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## STEREOISOMERISM IN 2-BENZALCYCLANONES1

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The base catalyzed aldol condensation of cyclanones with benzaldehyde is a well known reaction and leads to products to which the structure of trans-2-benzalcyclanones, e.g. I, has been ascribed. 3,4 Cornubert and coworkers3 assigned the structure of cis-2-benzalcyclanones, e.g. IV, to isomeric products resulting from acid catalyzed aldol condensation of several cyclanones with benzaldehyde. Previous work with benzalcyclanones4,5 made this structure assignment doubtful. Reinvestigation of this problem showed that the compound obtained in the acid catalyzed aldol condensation of 2-methylcyclohexanone or upon isomerization of I ( $\lambda$ 284 mm;  $\xi$  16,000) with acid actually possesses structure VI ( $\lambda_{max}$  234 mm; £ 9,000). Thus 2-benzyl-6-methyl-2-cyclohexenone (VI) upon ozonolysis yielded phenyl acetic and X-methyl glutaric acid. Sodium borohydride reduction of VI gave 2-benzyl-6-methyl-2-cyclohexenol reconvertible to VI with manganese dioxide but devoid of conjugative absorption in the ultraviolet. Ketone I, on the other hand, is reduced to 2-benzal-6-methylcyclohexanol (VIII) that absorbs at 244 mm; £ 19,000.

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It is noteworthy that these isomerizations proceed from the more to the less conjugated system and in five as well as in six-membered rings. Thus trans-2-benzal-5-methylcyclopentanone (III) is converted by acid to

I X=1, R=CH, R'=H

IV X=1, R=CH, R'=H

II X=1. R=R'=CaHa

V X=1, R=R'=C\_H

VII X=0

III X=O, R=CH, R'=H

its endocyclic unsaturated isomer VII. 6,6-Disubstituted-2-benzalcyclohexanones however are not isomerized under these conditions.

IV and V, the actual cis isomers of benzalketones I and II, 6 were obtained in good yield by light irradiation of solutions of I and II respec-IV (λ<sub>max</sub> 270 mμ; ξ 6,700) is converted by heat into I and by treatment with acid into VL The infrared spectra of cis benzal ketones IV and V indicate steric inhibition of orbital overlap (N Caro 1690 cm-1, intensity ratio  $\frac{\sqrt{C=C}}{\sqrt{C=O}} = 0.2-0.3$ ) with respect to the spectra of the <u>trans</u> isomers ( $V_{C=0}$  1668 cm<sup>-1</sup>, intensity ratio  $\frac{V C=C}{V C=0}$  1).

The trans isomer III, reported by J. P. Collman, J. Org. Chem. 26, 3162 (1961) to absorb at 270 mm actually absorbs at 292 mm but converts readily to the cis isomer VI which absorbs at 268 mm.